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Chemistry Books; 1.  $2\text{ClF}(\text{g}) + \text{O}_2(\text{g}) \leftrightarrow \text{Cl}_2\text{O}(\text{g}) + \text{F}_2\text{O}(\text{g})$

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$\Delta H = 167 \text{ kJ/mol rxn.}$

During the reaction above, the product yield can be increased by increasing the temperature of the reaction. Why is this effective? A. The reaction is endothermic; therefore adding heat will shift it to the right. B. Increasing the

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Periodicity Name \_\_\_\_\_ MULTIPLE  
CHOICE. Choose the one



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alternative that best completes the statement or answers the question. 1) Ham radio operators often broadcast on the 6-meter band. The frequency of this electromagnetic radiation is \_\_\_\_\_ MHz. A)50 B)20 C)2.0 D)200 E)500 2) What is the frequency of

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Tight ( $s^{-1}$ ) that has a wavelength  
...

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CHOICE. Choose the one

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alternative that best completes the statement or answers the question. 1) The gain of electrons by an element is called \_\_\_\_\_.  
A)oxidation B)sublimation  
C)reduction D)disproportionation  
E)fractionation 2) \_\_\_\_\_ is reduced in the following reaction:  $\text{Cr}_2\text{O}_7^{2-}$

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7 Atomic Structure And  
 $+ 6S^{2-} + 14H^+ \rightarrow 2Cr^{3+} +$   
 $3S_4O_6^{2-} + 7H_2O$  A)  $Cr_2O_7^{2-}$  ...

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to visit the AP Chemistry student page for exam information and exam practice.

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AP Chemistry Review | AP Practice

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Chemistry Books; 1. The  
subatomic particle with the most  
penetrating ability is. A. a beta  
particle. B. a neutron. C. a proton.

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D. an alpha particle. 2. Which of the following is expected to be a polar molecule? A.  $\text{PCl}_4\text{F}$ . B.  $\text{BF}_3$ . C.  $\text{CO}_2$ . D.  $\text{Si}(\text{CH}_3)_4$ . 3. All of the following ions have the same electron configuration except. A.  $\text{Rb}^+$ . B.  $\text{Se}^{2-}$  ...

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Sample Question 1 Ap chemistry  
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minutes to submit). Common  
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