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Lifewave Webinar 1.1

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Pure Substances and Mixtures, Elements \u0026amp; Compounds, Classification of Matter, Chemistry Examples,

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3.2.1/3.2.2 Describe the differences between elements, compounds and mixtures. States of Matter : Solid Liquid Gas 10 Amazing Experiments with Water Changes in the State of Matter - Science Experiments -

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Solid to Liquid, Liquid to Gas, Gas to Liquid Changes of States of Matter Conservation of Mass in Chemical Reactions The States of Matter - Changes of State - Solid Liquid Gas Plasma | Physical Science | Gen Chemistry States of Matter Understanding Atoms, elements, and molecules Part #1 (9min) Study Jams - Properties of Matter States of Matter and Changes of State - Science for Kids

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Chemistry: The Molecular Nature Of Matter And Change 7th ...  
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CHAPTER 9 SOLUTIONS MANUAL 35. Aqueous solutions of potassium iodide and silver nitrate are mixed, forming the precipitate silver iodide. chemical equation:  $KI(aq) + AgNO_3(aq) \rightarrow KNO_3(aq) + AgI(s)$

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AgI(s) complete ionic equation:  $K(aq) + I(aq) + 3AgNO_3(aq) \rightarrow KNO_3(aq) + AgI(s) + 3AgNO_3(aq)$

## Chemical Reactions

**Solution.** Physical: boiling and melting are physical changes. When water boils, no bonds are broken or formed. The change could be written:  $H_2O(l) \rightarrow H_2O(g)$  Chemical: The dark grey nail changes color to form an orange flaky substance (the rust); this must be a chemical change.

## 3.6: Changes in Matter: Physical and Chemical Changes ...

240 Chemistry: Matter and Change • Chapter 12 Solutions Manual  
CHAPTER 12 SOLUTIONS MANUAL 30. Compare and contrast sublimation and evaporation. In both processes, the substances become a vapor. During sublimation, the substance goes from the solid phase directly to the vapor phase. During evaporation, particles in a liquid gain enough

## States of Matter

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